Blocking amyloid assembly with chemical denaturants

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Amyloid fibrils

Protein aggregation is an alternative to monomeric folding

monomers \rightarrow aggregates \rightarrow amyloid fibrils

Amyloid fibrils:

- 1. Long unbranched nanostructures
- 2. Extremely stable, form irreversibly
- 3. Assembly timescale >> folding timescale
- 4. Generic structure available for <u>any</u> polypeptide sequence
- 5. Generally non-functional and cytotoxic, linked to a new class of diseases



Structure of amyloid fibrils

- Universal internal organization based on β-sheet in registry structure
- β-sheet is stabilized by hydrogen bond network
- β-sheets are laminated in layers



Amyloid fibril structures

First "real" conformation of peptides in fibrils (Serpell, *PNAS* 2005)

3D structure of the designed **KFFEAAAKKFFE** peptide

- Antiparallel β-sheets
- Layered brick-like arrangement of βsheets



$\ensuremath{\mathsf{A}\beta}$ peptides and Alzheimer's disease

- $A\beta$ peptides:
 - Natural product of cell proteolysis
 - Exist in a variety of lengths (39-42mers)
 - Form amyloid fibrils
- Aβ amyloid hypothesis for Alzheimer's disease (AD)
- Aβ oligomers are potent neurotoxins
- AD neurotoxic agent: Aβ fibrils or oligomers?
- Oligomers are possibly involved in other neurogenerative diseases



AEF RHDSGYEVHHQKLVFFAEDVGSNKSAIIGLMVGGVVIA

Molecular dynamics simulations of Alzheimer's Aß peptides



Hydrophobic

C-terminal



CHC

Hydrophilic

N-terminal

Assembly of A β amyloids



$A\beta_{16-22}$ peptide is a model amyloid system



MD simulations of $A\beta_{16-22}$ oligomers

- OPLS all-atom representation of three Aβ₁₆₋₂₂ peptides in water(+urea) (~4,000 atoms in 35Å x 35Å x 35Å unit cell)
- Probe the kinetics of oligomer assembly through multiple NVE MD trajectories starting with *random* initial structures (>150 *ns* simulation time)



Aβ16-22 monomers adopt random coil structure in water



Turning on interpeptide interactions...

Disordered oligomer is

stabilized by hydrophobic interactions



<u>Interpeptide interactions</u> drive an accumulation of β -structure No order in peptides' orientations





Assembly of ordered A β_{16-22} oligomers in water

Antiparallel orientation of peptides in *ordered oligomers*



Ordered oligomers are stabilized by *hydrophobic+electrostatic* side chain contacts



Structural ordering in A β 16-22 oligomers resembles fibril organization within ~ 10 ns

Question:

how do chemical denaturants affect oligomer assembly?

Motivation:

- 1. Tool for probing the mechanism of amyloid formation
- 2. Some amyloids (Ig light chain) are formed in the presence of urea
- 3. Implications for protein unfolding

- Urea enhances β -propensity in $A\beta_{16-22}$ peptides
- Distribution of monomer conformational states in urea (water)
 - ➤ 46% random coil (68%)
 - > 53% β-strand (29%)
 - \succ negligible amount of α -helix
- Conformational properties are static on a timescale of 10 ns



Effect of urea on the distribution of end-to-end distance $P(r_{1N})$



A β 16-22 monomers in 8M urea

What are the interactions that might explain enhanced β -propensity?



Enhanced β -propensity is due to urea-backbone hydrogen bonding

Urea solvates backbone better than water

- Relative gain factor in [U] over [W] is 3.3
- Average residence times in FSS indicate rapid exchange of solvent molecules

>
$$<\tau_{U}> = 14.1 \text{ ps}$$

> $<\tau_{w}> = 8.9 \text{ ps}$



Backbone FSS: [W]/[U]≈1.5

Enhanced β -propensity is due to urea-backbone hydrogen bonding

Solvation of backbone amide hydrogens:

- $<N_{U}>+<N_{W}>=2.2+3.1=5.3$
- Urea cross-bridges backbone
 better than water



 β -propensity is <u>**not**</u> caused by the solvation of side-chains

Water solvates *charged* side chains 3.5 better than urea 3.0 H(K) = O(U)H(K) = O(W)2.5 *(1) d(1)* Increase in [W]/[U] H(K) FSS: 50% 1.0 O(E) FSS: 30% 0.5 0.0 2 10 12 6 14 Ο 4 8 r, Å

 β -propensity is **<u>not</u>** caused by the solvation of side chains



$\mbox{A}\beta_{16\mbox{-}22}$ monomers in 8M WS urea

Modifying urea partial charges (Weerasinghe and Smith (WS))

5

2

 \bigcirc

g(r)

- O(U): 0.390 → -0.675
- H(U): +0.333 → +0.285

Effect of WS potentials:

- 1 further enhance β -propensity
- 2.improve backbone solvation with urea
 - •relative gain [U]/[W] = 5.5



8

r, Å

10

12

14

6

Conclusion:

urea-induced structural changes are electrostatic in origin

A β_{16-22} oligomer in aqueous 8M urea solution



8M urea *dissolves* $A\beta_{16-22}$ oligomers

- <ASA>: 2700 → 3200 A² in 11 ns
- <R^{cm}(t)>: a 50% increase

Comparison: in pure water <ASA(t)>≈const



- 8M urea dissolves $A\beta_{16-22}$ oligomers *irrespective* of
- urea model or initial conditions



Oligomer is disrupted within 11ns

Dynamics of solvation of peptides' backbones:

Urea "invades" into A β_{16-22} oligomers



Result: urea covers hydrophobic residues ([U] increases by 50%)

Urea *accelerates* α -helix $\rightarrow \beta$ -strand conformational transition



Summary of urea effect on A β_{16-22} oligomers

The impact of urea on $A\beta_{16-22}$ oligomers is two-fold:

- Destabilizes and disrupts $A\beta_{16-22}$ oligomers
- Accelerates and enhances β-structure formation

Through

- Hydrogen bonding to backbone amides
- Disruption of hydrophobic interactions

Prediction:

- 1. High [U] is likely to block amyloid formation
- 2. Moderate [U] may accelerate amyloid assembly

Outlook

Experiments on β -lactoglobulin



Outlook

MD simulations of $A\beta_{16-22}$ peptides:

- 1. Consistent with experiments (Prot. Sci. 11, 2417(2002))
- 2. Probes the mechanism of amyloid assembly
- 3. Applicable to other amyloidogenic polypeptides
- 4. Probes the mechanism of protein unfolding

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